## S4 ACEITEKA CHEMISTRY P3 CONFINDENTIAL

## Each student should be provided with the following

1 Burette <b>(50cm³)</b>
1 Pipette <b>(20 or 25cm³)</b>
2 Conical flasks
2 Beakers
1 Retort stand
6 Test tubes
1 Boiling tube
1 filter paper
90cm <sup>3</sup> of BA1
100cm <sup>3</sup> ofBA2
2g of Q
BA1 is made by dissolving 4g of sodium hydroxide in water to make 1dm³ of solution
BA2 is 0.105M hydrochloric acid
Q is a mixture of ZnCO <sub>3</sub> and CuCO <sub>3</sub> in the ratio 1:2