

BEGINNING OF TERM II EXAMINATIONS 2018

PHYSICS P510/1

INSTRUCTIONS:

- Attempt not more than one question from each section.
- Each section question begins on a new page
- All questions carry equal marks.

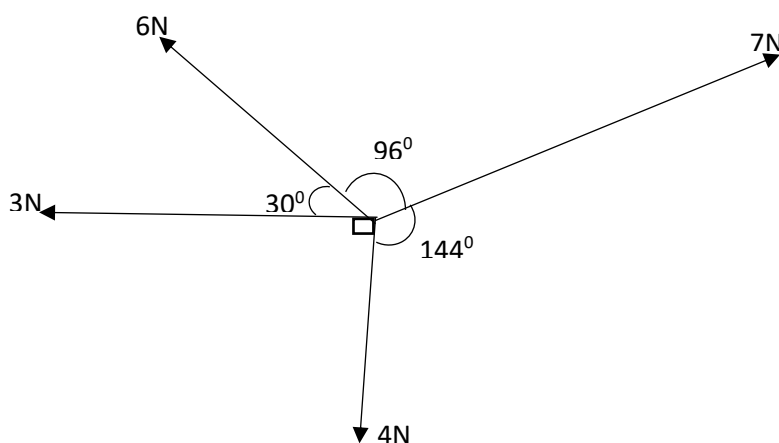
Time allowed: 1 hour 30 minutes

Where necessary assume the following;

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| ▪ Electronic charge, (e) | $= 1.6 \times 10^{-19} \text{C}$ |
| ▪ Acceleration due to gravity, (g) | $= 9.81 \text{ ms}^{-2}$ |
| ▪ Electron mass, (m_e) | $= 9.11 \times 10^{-31} \text{kg}$ |
| ▪ Mass of the earth, (M_E) | $= 6.0 \times 10^{24} \text{kg}$ |
| ▪ Plank's constant, (h) | $= 6.63 \times 10^{-34} \text{ Js}$ |
| ▪ Stefan's constant, (σ) | $= 5.7 \times 10^{-8} \text{ Wm}^{-2} \text{ K}^{-4}$ |
| ▪ Radius of earth (R_e) | $= 6.4 \times 10^6 \text{m}$ |
| ▪ Radius of the sun (R_s) | $= 7.0 \times 10^8 \text{ m}$ |
| ▪ Speed of light in a vacuum (c) | $= 3.0 \times 10^8 \text{m/s}$ |
| ▪ Specific latent of fusion of ice (L_f) | $= 3.36 \times 10^5 \text{Jkg}^{-1}$ |
| ▪ Specific heat capacity of water (C) | $= 4200 \text{Jkg}^{-1} \text{K}^{-1}$ |
| ▪ Universal Gravitation constant, (G) | $= 6.67 \times 10^{-11} \text{ Nm}^2 \text{ kg}^{-2}$ |
| ▪ Avogadro's number (N_A) | $= 6.02 \times 10^{23} \text{ mol}^{-1}$ |
| ▪ Density of water (ρ) | $= 10^3 \text{kgm}^{-3}$ |
| ▪ Surface tension of water () | $= 7.2 \times 10^{-2} \text{ Nm}^{-1}$ |
| ▪ Charge to mass ratio (e/m) | $= 1.8 \times 10^{11} \text{ Ckg}^{-1}$ |
| ▪ Density of copper | $= 8.93 \times 10^3 \text{ kgm}^{-3}$ |
| ▪ Specific heat capacity of copper | $= 3.70 \times 10^2 \text{ Jkg}^{-1} \text{ K}^{-1}$ |

SECTION A

1. a) Distinguish between fundamental and derived physical quantities giving two examples of each. (4 marks)
 - b) What is meant by the terms dimensions of a physical quantity and dimension index. (2 marks)
 - c) Derive the dimensions of the following:
 - i) Pressure (3 marks)
 - ii) Specific heat capacity (3 marks)
 - iii) Moment of a force (3 marks)
 - iv) tensile stress (2 marks)
 - d) The mass of the largest stone that can be moved by a flowing river is given by $m = \frac{KPV^6}{g^5}$ where V is the velocity of water, P is it's density, g is the acceleration due to gravity and K is a dimensionless constant. Prove that the equation is dimensionally consistent.
2. a) Define scalar and vector quantities. (2 marks)
 - b) Identify the following quantities as vectors or scalars. Momentum, Pressure, temperature, Work, acceleration, electric intensity, electric potential charge. (4 marks)
 - c) i) What is meant by resultant vector? (1 mark)
 - ii) The figure below shows four forces acting on a particle B of mass 5 kg initially at rest. Find the velocity at a point of 20 m from its initial position. (7 marks)



- d) i) Define the terms uniform velocity and acceleration. (2 marks)
 - ii) Give two examples of uniformly accelerated motion. (2 marks)
 - iii) How does the term instantaneous velocity differs from the term average velocity. (2 marks)

SECTION B

- 4.a) i) State the desirable properties a material must have to be used as a thermometric substance. (02 marks)
- ii) Thermometers based on different thermometric properties have cases when they agree and when they may not. Explain why and when this can happen. (03 marks)
- b) i) Describe how a simple constant volume gas thermometer can be used to establish a Celsius scale of temperature. (04 marks)
- ii) State any reasons why one may use or not use a constant volume gas thermometer (03 marks)
- iii) When the gas bulb of a constant volume gas thermometer was immersed in crushed ice, the difference in mercury level indicated 5 cm. The same bulb was placed above steam from boiling water and the difference in mercury levels was 15 cm. If the unknown temperature calculated was 80°C , what was the difference in the mercury levels at this unknown temperature. (04 marks)
- c) i) Which thermometric property do pyrometers use? (01 mark)
- ii) Name the two types of pyrometers that measures temperature (02 marks)
- iii) When would one opt for pyrometers during measurement of temperature. (01 mark)

SECTION C

- 5.a) Define the following terms as applied to radioactivity
- i) Nuclear number (1 mark)
- ii) Decay constant (1 mark)
- iii) Half life (1 mark)
- b) Show that the relationship between half life, $t_{\frac{1}{2}}$ and the decay constant, λ is
- $$t_{\frac{1}{2}} = \frac{0.693}{\lambda} \quad (3 \text{ marks})$$
- c) A radioactive material initially has an activity of 3.5×10^3 per second. If the nuclide has a half life of 30 days. Calculate,
- (i) the initial number of atoms present. (2 marks)
- (ii) the activity after 20 days (4 marks)
- d) Describe the structure and mode of action of a cloud chamber. (5 marks)
- e) Explain how carbon dating can be used in determining the age of a plant remain. (3 marks)

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