

**P510/1**  
**PHYSICS**  
**Paper 1**  
**Nov 2015**  
**2½ HOURS**

**Uganda Advanced Certificate of Education**  
**S.5 PHYSICS**  
**Paper 1**  
**2hours 30minutes**

**INSTRUCTIONS TO CANDIDATES:**

Attempt four questions but not more than two from each of the sections A and B.

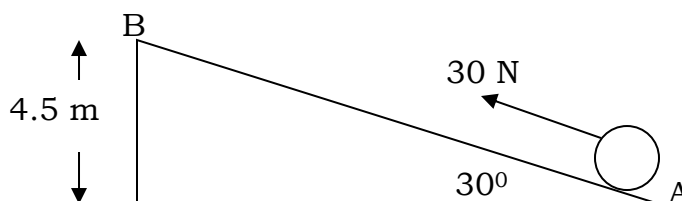
Non-programmable scientific electronic calculators may be used

The following values of physical constants may be used.

Acceleration due to gravity	$g$	=	$9.8\text{ms}^{-2}$
Specific heat capacity of water		=	$4200\text{Jkg}^{-1}\text{K}^{-1}$
Specific heat capacity of copper		=	$400\text{Jkg}^{-1}\text{K}^{-1}$
Velocity of light	$C$	=	$3.0 \times 10^8\text{ms}^{-1}$
Avogadro's number	$N_A$	=	$6.02 \times 10^{23}$

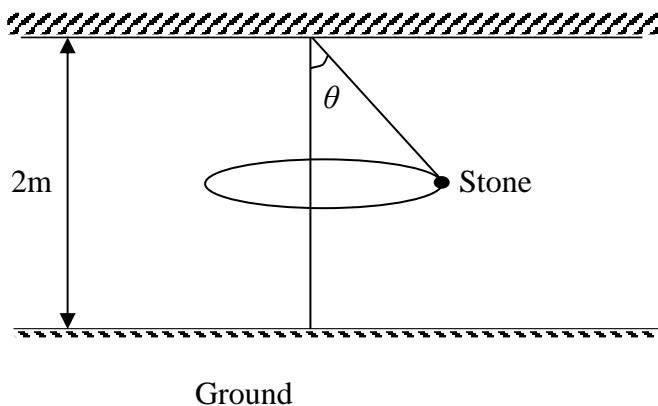
**SECTION A**

1. (a) Define the term work and state its S.I unit. (01 marks)
- (b) A particle of mass 3.0kg is pulled by a constant force of 30N along an inclined plane AB with height 4.5 m inclined at an angle  $30^\circ$  to the horizontal as shown below.



- (i) Calculate the work done by the force to move the particle from A to B. (03 marks)
- (ii) The acceleration of the particle. (03 marks)
- (c) (i) State Newton's laws of motion. (03 marks)
- (ii) Define the term impulse. (01 mark)
- (d) Explain why a heavy truck takes longer to brake than a light truck when both are moving with the same velocity. (03 marks)
- (e) (i) What is meant by the term perfectly inelastic collision? (01 mark)
- (ii) A bullet of mass 50g is fired from a gun at  $250 \text{ ms}^{-1}$  and hits a block of wood of mass 2kg which is suspended by a light vertical string 2m long. Calculate the maximum angle the string makes with the vertical if the bullet got embedded in the wood after collision. (04marks)
- (iii) State two (2) factors on which the angle of swing depends on. (02 marks)

- 2(a) (i) Derive an expression centripetal acceleration of a body moving in a circular path of radius  $r$ . (03 marks)
- (ii) Explain why it is necessary for a bicycle rider moving around a circular path to lean towards the center of the path. (03 marks)
- (iii) Derive the expression for the angle inclination to the horizontal necessary for a rider moving a round a circular track of radius  $r$  without skidding at a speed  $v$ , in terms of  $g$ ,  $r$ , and  $v$ . (04 marks)
- (ii) A uniform beam AB of mass 4.00kg and length 3m is pivoted at a point P and kept in equilibrium by two masses of 20.75kg and 1.50kg at points A and B respectively. What is the distance of P from A? (04 marks)
- (ii) Describe how the centre of gravity of an irregular piece of card board may be determined. (4 marks)
- (b) (i) State any two conditions necessary for a rigid body to be in equilibrium. (02 marks)
3. (a) (i) Define moment of a force and state its units. (02 marks)
- (ii) Outline any four applications of principle of moments. (02 marks)
- (d) The period  $T$  of oscillation of a simple pendulum depends on length  $l$ , of the string and acceleration due to gravity,  $g$  as shown in the formula  $T = 2\pi \sqrt{\frac{l}{g}}$  Show that the equation is dimensionally consistent. (04 marks)
- c. Explain any two applications of dimensions of a physical quantity. (02 marks)
- (b) A stone of mass 0.5kg is tied to one end of a string 1m long. The point of suspension of the string is 2m above the ground. The stone is whirled in a horizontal circle with increasing angular velocity.



The string will break when the tension in it is 12.5N and angle  $\theta$  is a maximum.

- (i) Calculate the maximum value of  $\theta$  (03 marks)
- (ii) How far from point **G** will the stone hit the ground? (04 marks)
- (iii) What will be the velocity of the stone when it hits the ground? (03 marks)

### SECTION B

- 4 a.i) Define specific latent heat of vaporization. (01 mark)
- ii) Explain why specific latent heat of vaporization is always greater than specific latent heat of fusion for the same substance using the molecular theory. (02 marks)
- b) Describe an experiment to determine the specific latent heat of vaporization of water using method of mixtures. (06 marks)
- c) A calorimeter of negligible heat capacity is filled with 1.5 kg of molten substance. The rate of cooling just before solidification begins is  $8.4 \text{ K s}^{-1}$  and complete solidification takes 25 minutes. Given that the total heat lost in the process was 236100J
- (i) Sketch the cooling curve. (01 mark)
  - (ii) If the specific latent heat of fusion of the substance is  $31400 \text{ JK g}^{-1}$ , calculate its specific heat capacity. (04 marks)
- d) (i) State Newton's law of cooling. (01 mark)
- (ii) Describe an experiment to verify Newton's law of cooling. (05 marks)
5. (a) (i) What is meant by a thermometric property of a substance? (01 mark)
- (ii) Give any two examples of these properties. (01 mark)
- (i) Define the term fixed point. (01 mark)
- (b) (i) Define the scale of temperature on a constant volume gas thermometer. (01 mark)
- (ii) With the aid of a labeled diagram, describe the mode of operation of a practical constant volume gas thermometer. (04 marks)
- (iii) State the corrections necessary in the thermometer in b(i) above. (02 marks)
- (c) Explain using the kinetic theory why temperature remains constant during change of phase. (03 marks)
- (d) State any two effects of heat on substances. (02 marks)

- (e) (i) Explain briefly what is meant by absolute zero temperature. (02 marks)
- (ii) A piece of material of mass 500g at a temperature of  $90^{\circ}\text{C}$  is carefully dropped in a copper calorimeter containing 450g of water at  $25^{\circ}\text{C}$ . The set-up reaches a temperature of  $35^{\circ}\text{C}$ .
- Ignoring any heat losses calculate the specific heat capacity of the material. (03 marks)
6. (a) (i) State assumptions made in the derivation of the kinetic theory expression for the pressure of an ideal gas? (04 marks)
- (ii) Show that the pressure  $\mathbf{P}$  exerted by an ideal gas of density  $\rho$  is given by  $\mathbf{P} = \frac{1}{3} \rho \mathbf{c}^2$  where  $\mathbf{c}$  is the mean square speed of the gas molecules. (07marks)
- (ii) State any two ways in which real gases differ from ideal gases. (02 marks)
- (b) A cylinder contains  $0.25 \text{ m}^3$  of a gas at a pressure of  $2.0 \times 10^6 \text{ Pa}$  and a temperature of  $17^{\circ}\text{C}$ . Calculate
- (i) the number of moles of the gas in the cylinder. (02 marks)
- (ii) the pressure that the gas would exert at a temperature of  $37^{\circ}\text{C}$  if its volume is kept constant. Distinguish between saturated and unsaturated vapours. (02 marks)
- (c) Draw sketch graphs to show, for a saturated vapour, the variation of:
- (i) Pressure with volume, temperature remaining constant. (01 mark)
- (ii) Explain the shape of the curve in (i) above, using the kinetic Theory of matter. (02 marks)

**END**